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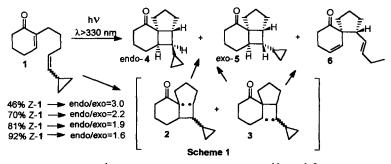
## The Order of First Bond Formation in the Intramolecular Photocycloaddition of 2-(4'-pentenyl)-cyclohex-2-enones

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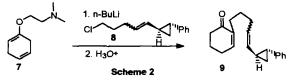
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**Abstract:** Complete trapping of 1,4-diradical intermediates, formed in the photocycloaddition of 9, provide direct evidence that first bond formation takes place at either  $C(\alpha)$  or  $C(\beta)$  positions of the cyclic enone. No cleavage of the 1,4-diradical 10 or *endo*-diradical intermediates 11 and 12 could be detected. However, partial cleavage of the *exo*-diradical intermediates 13 and 14 was measured. Unusual long-range aromatic radical substitution was obtained in a competing rate with the cyclopropyl carbinyl radical rearrangement. © 1997 Elsevier Science Ltd.

The intramolecular photocycloaddition of alkenes to cyclic enones of type 1, has been investigated by Becker and co-workers.<sup>1</sup> The regioselectivity of this reaction, which is related to the preferred position of the first bond formation and the validity of the "rule of five"<sup>2</sup> is still under active investigation in several research groups.<sup>3</sup> Becker has shown that irradiation<sup>1b</sup> of compounds 1 (Z/E=1.5) at 36°C afforded two isomers of [2+2]-photoproducts 4 and 5 (56%) and a trapping product 6 (37%). The *endo/exo* ratio of the [2+2]-photoproducts was strongly affected by the ratio of the Z/E isomers at the alkenyl side chain in the starting material 1. Based on this, it was concluded that [2+2]-photoproducts are formed *via* both diradical intermediates 2 and 3.



Recently we have demonstrated<sup>4</sup> that the fast rearrangement  $(3x10^{11} \text{ sec}^{-1})^5$  of 2'-phenylcyclopropyl carbinyl radical to homoallyl radical afforded complete trapping of 1,4-diradical intermediates. Applying this method in the photocycloaddition of 1 was expected to afford complete trapping of the 1,4-diradical intermediates of type 3. We report here trapping experiments which revealed direct evidence that first bond formation in the photocycloaddition of 9 takes place mainly at the C( $\alpha$ ) position of the cyclic enone and a smaller percentage (11%-22%) at the C( $\beta$ ) position.



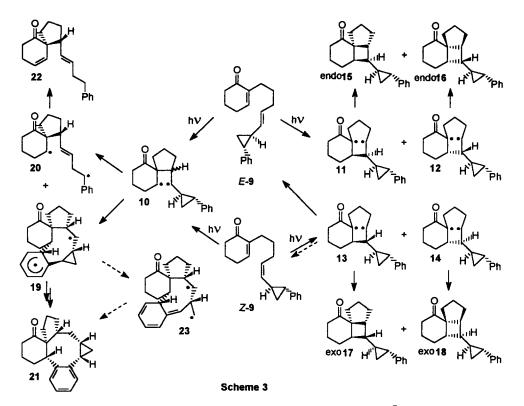
Compounds 9 were prepared by coupling of Z/E isomeric mixture of chiral cyclopropanes<sup>4</sup> 8 with 7, following previously reported procedure.<sup>1a</sup> The Z-9 and E-9 isomers were separated by column chromatography. Irradiation studies were carried out under the usual conditions<sup>6</sup> in the temperature range of -55°C to +40°C. The results are summarized in table 1. Based on previous studies,<sup>1</sup> only five diradical intermediates 10-14 should be considered after first bond formation of 9 under these irradiation conditions (scheme 3). Mixture of *endo*-15+16 and *exo*-17+18, are expected *via* diradical 10, whereas complete trapping of diradical 10 and selective formation of the [2+2] photoproducts *via* intermediates 11-14 (obtained by first bond formation at the C( $\beta$ ) position of the enone) and no cleavage of the diradical intermediates during the irradiation are essential features for obtaining *selective formation* of *endo*-15+16 or *exo*-17+18 photoproducts upon irradiation of *E*-9 or *Z*-9 respectively.

Entry	Comp.	Temp.	25	22	26	21	27	<i>endo-</i> 15+16	<i>exo-</i> 17+18
1	<i>E-9</i>	- 55 °C	4.6	54.5	1.7	17.1	4.8	19.3	
2	<i>E-</i> 9	0 °C	4.3	53.4	2.6	12.6	7.0	20.1	
3	E-9	+ 25 °C	3.8	49.4	4.3	9.7	11.5	21.2	
4	<i>E-</i> 9	+ 40 °C	3.9	47.8	4.2	10.3	11.8	21.9	
5	Z-9	- 55 °C	1.8	64.9		19.7	2.5		11.0
6	Z-9	0 °C	2.3	62.6	1.5	15.1	3.5	2.5	12.5
7	Z-9	+ 25 °C	2.7	61.1	2.2	12.5	5.1	3.1	13.4
8	Z-9	+ 40 °C	2.5	57.1	2.6	12.6	7.2	4.8	13.1

Table 1	1:	Irradiation	of <i>E-</i> 9	and	Z-9.
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Irradiation of *E*-9 at -55°C resulted in selective formation of the *endo*-[2+2] photoproducts 15+16 (19.3% total) with no detectable amount (GC) of the corresponding *exo*-17+18 isomers. However, irradiation of *Z*-9 at -55°C afforded *exo*-17+18 (11%) with no detectable amount of *endo*-15+16, along with the same trapping products 21 (19.7%) and 22 (64.9%). Compounds 15-18, 21 and 22 were separated and their structures determined by NMR.<sup>7</sup> The structures of the isomeric products (GC-MS) 25-27 (ca. 4 - 13%) were not determined. The fact that undefined products 25-27 obtained in the irradiation of both photosubstrates suggests their formation *via* diradical 10. The obtained regioselectivity is strongly in the favor of first bond formation at the  $C(\alpha)$  position of the cyclic enone, following the "role of five"<sup>2</sup>.

Interestingly, Irradiation of E-9 at different temperatures afforded the endo-[2+2] photoproducts (15+16) with no detectable amount of the corresponding exo-17+18 or  $E \rightarrow Z$  geometrical isomerization of the alkene in the starting material during the irradiation. However,  $Z \rightarrow E$  geometrical isomerization was obtained upon irradiation of Z-9 at higher temperatures along with formation of endo-[2+2] photoproducts (15+16) which is attributed to this isomerization. The observed endo/exo selectivity in the irradiation of E-9 at the temperature range of -55°C - +40°C and Z-9 at -55°C, clearly indicate that [2+2] photoadducts were exclusively formed via diradical intermediates 11-14, with complete trapping of the diradical intermediate 10. Similar selectivity found in the irradiation of Z-9 at higher temperatures (entries 6-8). The obtained endo-[2+2] photoproducts (15+16) is attributed to the ca. 5% detected geometrical isomerization of the alkenyl side chain in the starting material, obtained via fragmentation of the exo-diradicals 13 and 14 during the irradiation.



The structure of the unexpected trapping product 21 was determined by NMR.<sup>7</sup> The NOE enhancements obtained between the methyne protons, summarized in figure 1, allowed determination of the relative configuration at the stereogenic centers, further support of the structure was obtained by its X-ray structure analysis<sup>8</sup>. The formation of this product is proposed *via* diradical 19, obtained by unusual long-range intramolecular radical aromatic substitution at the *ortho*-position of the phenyl substituent, <sup>3b,9</sup> followed, presumably, by 1,6-hydrogen abstraction then photochemical isomerization of the cyclopropyl unit.<sup>10</sup> However, selective *exo*-ring opening of the cyclopropyl carbinyl radical<sup>11</sup> in 19, to its corresponding homoallyl radical 23 then recyclization to the *cis*-cyclopropyl diradical intermediate followed by 1,6-hydrogen abstraction to afford 21 could also be proposed.

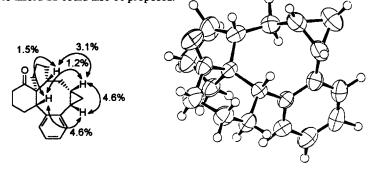


Figure 1: X-ray structure and selected NOE enhancements of compound 21

In summary, We present for the first time, direct evidence for the formation of diradical 10 and sevenmember ring diradical intermediates (11-14) in the intramolecular photocycloaddition of alkenes to cyclohexenones of type 9. The regioselectivity was accurately and unambiguously determined *via* complete trapping of diradical 10. The geometrical isomerization in photosubstrate Z-9 is attributed to the fragmentation of the *exo*-diradical intermediates 13 and 14 which is consistent with the low percentage (<10%) isomerization previously reported on the irradiation of related systems.<sup>1a</sup>

## Acknowledgment

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## **References and Notes:**

- 1. (a) Becker, D.; Haddad, N Tetrahedron 1993, 49, 947-964; (b) Becker, D.; Denekamp, C.; Haddad, N. Tetrahedron Lett. 1992, 33, 827-830.
- (a) Hammond, G. S.; Liu, R. S. J. Am. Chem. Soc., 1967, 89, 4930; (b) Carlough, K. H.; Srinivasan, R. J. Am. Chem. Soc., 1967, 89, 4932.
- (a) Maradyn, D. J.; Weedon, A. C. J. Am. Chem. Soc., 1995, 117, 5359; (b) Caldwell, R. A.; Hrneir, D. C.; Munoz, T. Jr.; Unett, D. J. J. Am. Chem. Soc. 1996, 118, 8741; (c) Broeker, J. L.; Eksterowicz, J. E.; Belk, A. J.; Houk, K. N. J. Am. Chem. Soc., 1995, 117, 5359; (d) Haddad, N. The Chemistry of Functional Groups, Supp. A3, S. Patai Ed., Wiley, 1997, 641.
- 4. Becker, D.; Galili, N.; Haddad, N. Tetrahedron Lett., 1996, 37, 8941-8944.
- 5. (a) Newcomb, M.; Johnson, C. C.; Manek, M. B.; Varick, T. R. J. Am. Chem. Soc., 1992, 114, 10915.
- 6. 80-W Hanau mercury vapor lamp (Q-81) was used for irradiations via a Uranium glass filter ( $\lambda > 330$ ). All irradiations were carried out in 18 mL of hexane as solvent under nitrogen atmosphere. The concentrations always kept below 0.05 M and the reactions followed by TLC or GC.
- All new compounds were characterized by full spectroscopic data, yields refer to chromatographed materials with purity of >95%. The relative stereochemical relationship of the stereogenic centers was determined by NOE-difference. The location of these protons determined by COSY-45, XH-CORR, JMOD-XH methods and was supported by NOE experiments. For determination of similar structures by NMR see ref. 1a and: Becker, D.; Haddad, N. Isr. J. Chem. 1989, 29, 303.
- X-ray analysis of 21: Empirical formula C<sub>20</sub>H<sub>24</sub>O, F.W. 280.4, T = 293K, lattice monoclinic, space group P2<sub>1/n</sub>, a = 13.632(6), b = 9.088(5), c = 25.465(12) Å, β = 97.37(5)°, V = 3129.6 Å<sup>3</sup>, Z = 8, F(000) = 1216, Dx = 1.191 g.cm<sup>-3</sup>, radiation Mo Kα, λ =0.7107 Å, μ = 0.071 mm<sup>-1</sup>, θ range for data collection 2-25°, index ranges -16≤h≤16, 0≤k≤0, 0≤1≤30, No. of unique reflections 5564, structure solved by SHELXS86<sup>(a)</sup>, and refined by SHELXL76<sup>(b)</sup> programs, refinement method full matrix least squares on F<sup>2</sup>, 2092 reflections for which Fo≥3σ(Fo), No. of parameters 571, R = 0.077, wR2 = 0.118, largest diff. peak 0.183 eÅ<sup>-3</sup>. (a) Sheldrick, G. M. Acta Crystallogr. 1990, A46, 467; (b) Sheldrick, G. M. (1976) SHELX76, program for crystal structure determination. University of Cambridge, England.
- 9. For recent examples of 1,6-radical aromatic substitution see ref. 3b and: Caldwell, R. A.; Diaz, J. F.; Hrneir, D. C.; Unett, D. J. J. Am. Chem. Soc. 1994, 116, 8138.
- 10. Koch-Pomeranz, U.; Schmid, H.; Hansen, H. J. Helv. Chim. Acta 1977, 60, 768.
- 11. Batey, R. A.; Grice, P.; Harling, J. D.; Motherwell, W. B.; Rzepa, H. S. J. Chem. Soc., Chem. Commun. 1992, 942.

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